## **Industrial Gases and Chemicals Questions**

- 1. Discuss a method of manufacturing NaOH.
- 2. Bleaching powder loses its bleaching property when kept in an open bottle for a long time.
- 3. Describe the iodometric method for estimating available chlorine in the given sample of bleaching powder.
- 4. Discuss the manufacturing process of acetylene gas and specify its three uses.
- 5. Define acid mine drainage.
- 6. 10 g of bleaching powder on reaction with KI required 50 mL of 2*N* hypo solution. What is the % of bleaching powder?
- 7. In each sample of bleaching powder, the percentage of available chlorine is 49. What volume of chlorine obtained if 10 g of the sample is treated with HCl at *N.T.P.*?
- 8. Define  $O_2$  sag curve.
- 9. What is oleum? Explain the contact process for manufacturing sulphuric acid. What are the main applications of sulphuric acid?
- 10. Write short notes on manufacture and use of bleaching powder.
- 11. How is hypo is manufactured? What are the main applications of hypo and what precautions should be kept in mind while using it?
- 12. Define cryogenic air separation.
- 13. Manufacture and uses of nitrogen.
- 14. What are the hazards associated with NaOH?
- 15. List some medicinal uses of KMnO<sub>4</sub>. How it is prepared?
- 16. Give a method for producing  $K_2Cr_2O_7$ . How is it hazardous?
- 17. Except for fractional distillation what other method may be used to manufacture nitrogen gas.
- 18. Liquid nitrogen needs careful handling and storage, why?
- 19. Explain the charcoal method of separation of neon gas from the mixture of inert gases.
- 20. List three applications of neon gas.
- 21. Give the hazards associated with argon gas.
- 22. List some uses of argon.
- 23. Name the different sources of  $SO_2$  in environment. How can  $SO_2$  emission can be controlled?
- 24. SO<sub>2</sub> is hazardous to vegetation as well as animal. Comment.
- 25. Explain how borax may be used as
  - (i) Water softener
  - (ii) Buffer
  - (iii) In medicines
- 26. How is borax manufactured?
- 27. Borax is a polyborate, explain with its structure.
- 28. What are the advantages and disadvantages of Diaphragm cell electrolysis method of chlorine production? How does these disadvantages take care of in membrane cell electrolysis method?
- 29. List different applications of chlorine. With equation how chlorine is used to disinfect water?
- 30. Give method of storing chlorine safety and write the hazards associated with it.