## **Air Pollution**

- 1. Differentiate between the Primary and secondary pollutants.
- 2. Define the following with suitable examples/reactions.
  - (i) Acid rain
  - (ii) Octane number
  - (iii) Inversion
  - (iv) ALBEDO
  - (v) Oxygen enrichment process
- 3. Give a labelled self-explanatory diagram of biogeochemical cycle of sulphur.
- 4. Explain the consequences of global warming briefly.
- 5. Automobiles are the major source of CO, hydrocarbons and  $NO_x$  mixture. How can this be converted to less harmful pollutants? Discuss in detail.
- 6. Give an account of inorganic and organic particulate matter present in the atmosphere. Discuss a method of controlling emission of particulate matter.
- 7. Give the mechanism involved in PAN formation during photochemical smog.
- 8. What are the major sources of CO? Give the method of controlling the amount of CO in the atmosphere.
- 9. What are the major sources and sink of CO in atmosphere? Discuss a method of measuring CO in an air sample.
- 10. Write shot notes on role of catalytic convertors in automobiles.
- 11. Draw a labelled diagram of biogeochemical cycle of carbon.
- 12. What are the major sources and sinks of different  $NO_x$ ? Give a method of estimating the amount of  $NO_x$  in the air sample.
- 13. Name four different greenhouse gases. How are they affecting global warming?
- 14. How does catalytic convertor in an automobile reduce air pollution?
- 15. Illustrate the different regions of atmosphere, specify the different chemical species and biota present in the different regions.
- 16. Write short notes on the following:
  - (i) Photochemical smog
  - (ii) Method of controlling particulate matter in air

## Water Pollution

- 17. What are the pollutants present in the industrial effluent of petroleum industry? How can these pollutants be taken care off?
- 18. Define DO. Discuss the method for the estimation of DO in water sample.
- 19. What is BOD? Give the difference between carbonaceous BOD and nitrification BOD. 5 day BOD of some waste water was found to be 200 mg/L. If the reaction rate constant K = 0.22/day, find ultimate BOD.
- 20. List the pollutants present in waste water of 'electroplating' and 'dairy' industries. Suggest a treatment method for the same
- 21. Give three properties of water that make it essential for life.
- 22. What is eutrophication and how can it be controlled? What is Liebig's Law? How can Liebig's law be explained with eutrophication as an example?
- 23. Name the pollutants present in the effluents of each tannery and textile industries. Suggest a method for the treatment for the same.

- 24. Draw a labelled diagram of a thermally stratified water body specifying the temperature, chemical species and biota existing in each region.
- 25. Discuss the role of ion exchange method for the water purification.
- 26. How is sludge disposed during secondary treatment of water, explain with reactions.
- 27. How do fertilizers pollute a water body? How can this pollution be controlled? Give Liebig's law of minimum and explain its significance.
- 28. Write shot notes on the disposal methods of different nuclear wastes.
- 29. Discuss aerobic digestion process applied during treatment of water.
- 30. Write short notes on the following:

(ii) Disposal of sludge, (ii) BOD and COD, (iii) Oil spills as water pollutant, (iv) Tertiary treatment of water