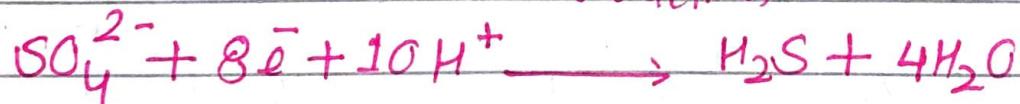
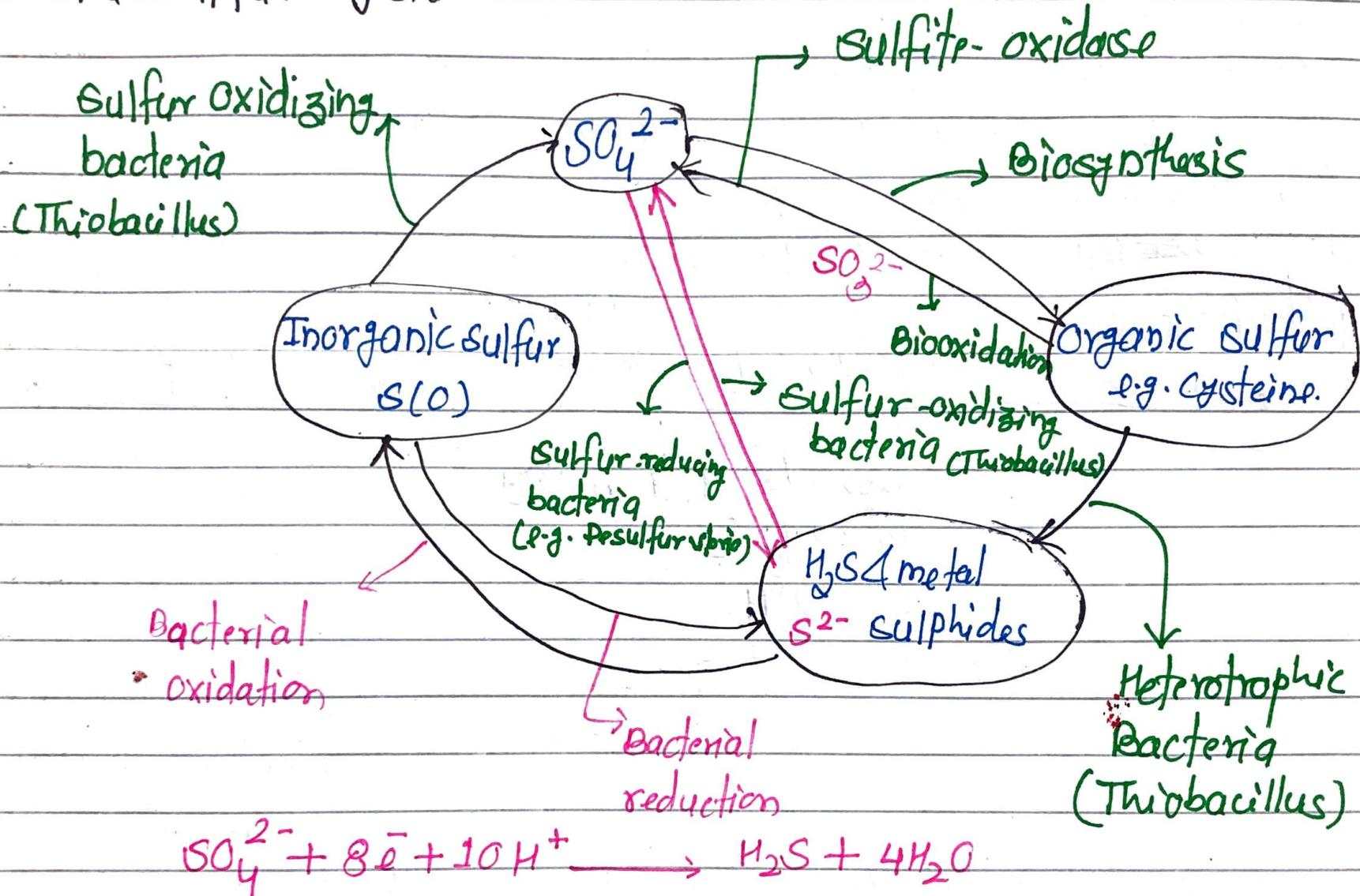


# The sulfur Cycle:



# Atmosphere

- The atmosphere is a protective blanket that nurtures life on the Earth and protects it from the hostile environment of outer space.
- The atmosphere is a key constituent of Earth's natural capital.
- Atmosphere is a source of

$\rightarrow \text{CO}_2$   
 $\rightarrow \text{O}_2$   
 $\rightarrow \text{N}_2$   
 $\rightarrow \text{H}_2\text{O}$   
 $\rightarrow \text{other gases}$

## Role of Atmosphere

- |                                    |  |  |   |
|------------------------------------|--|--|---|
| Transport water from ocean to land | Dumping grounds for many pollutant materials.<br>e.g. $\text{SO}_2$ , chlorofluorocarbons etc. | As a protective shield<br>• Absorbs most of the cosmic rays from outer space<br>• Filtering out damaging UV radiations (below 300nm) | Stabilizes the Earth's temperature.<br>• Atmosphere reabsorbs much of the IR radiation by which absorbed solar energy is re-emitted to space.<br>• Atmosphere preventing the tremendous temp. extremes that occur on planets and moons lacking substantial atmospheres. |
|------------------------------------|--|--|---|

# Components of Atmosphere

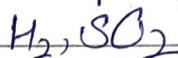
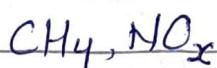
## Major Components

- $N_2$  78.08%
- $O_2$  20.95%

## Minor

- Ar 0.934%
- CO 0.040%
- Ne  $1.818 \times 10^{-3}\%$
- He  $5.24 \times 10^{-4}\%$
- Kr  $1.14 \times 10^{-4}\%$
- Xe  $8.7 \times 10^{-6}\%$

## Trace



etc.

## Water by

volume.

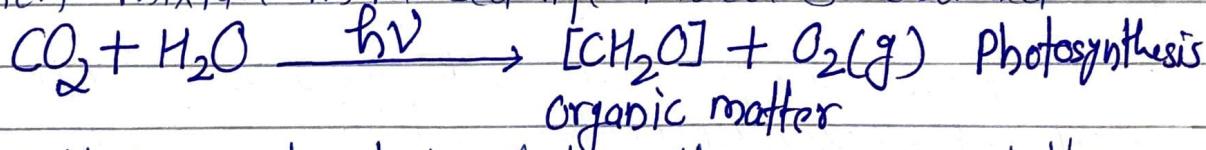
0.1 - 5%

Normal range

1% - 3%

## Chemical and Biochemical in Evolution of the Atmosphere:

- Approximately 3.5 billion years ago when the first primitive life molecules were, the atmosphere was probably free of  $O_2$  and consisted of a variety of gases such as  $CO_2$ ,  $H_2O$  vapour and probably  $CH_4$ ,  $NH_3$  and  $H_2$ .
- The atmosphere was bombarded by intense bond breaking UV light which along with lightning and radiation from radionuclides, provided the energy to bring about chemical reactions that resulted in the production of relatively complicated molecules, including even amino acids and sugars. From the rich mixture in the sea life molecules evolved.



- The oxygen initially produced by photosynthesis was probably quite toxic to primitive life forms. However much of this  $O_2$  was converted to  $Fe_2O_3$  by Reaction with soluble  $Fe(II)$



# Major Regions of the Atmosphere

