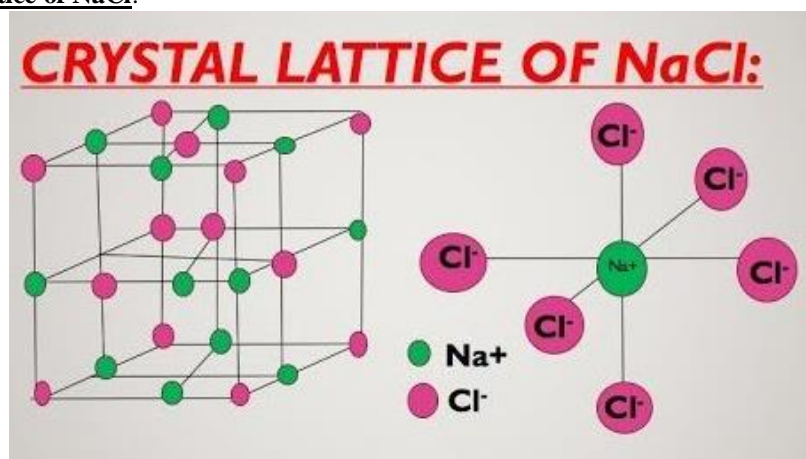


Home Assignment on the Topic Solid State Chemistry

1. What are the differences between crystalline and amorphous solids?
2. When a certain crystal was studied by Bragg's method using X-rays of wavelength 229 pm, first order X-ray reflection was observed at an angle of $23^{\circ}20'$:
 - (i) What is corresponding inter-planar spacing?
 - (ii) When another X-ray source was used, a reflection was observed at $15^{\circ}26'$. What was the wavelength of these X-rays?
3. Write short notes on the following:
 - (i) Law of rational indices
 - (ii) Law of constancy of interfacial angle
 - (iii) Schottky defect and Frenkel defect.
 - (iv) Unit cell and space lattice
 - (v) Law of crystallography
 - (vi) Liquid crystal
4. Determine the value of d_{hkl} in terms of the cell constants for the orthorhombic and tetragonal unit cells.
5. What are Miller indices? How are they determined?
6. Derive Bragg's equation of X-ray crystallography.
7. How many independent Bravais lattices are there in a cubic crystal system? Name them. Using a diagram show 100 and 110 planes in a primitive cubic lattice.
8. Determine the interplanar spacing between the (221) planes of a cubic lattice of length 450 pm.
9. The distance between two consecutive (110) planes of a crystal is 1.678×10^{-10} m. What will be the glancing angle for an X-ray of wavelength 0.65×10^{-10} m incident on the plane for first order reflection?

Crystal Lattice of NaCl:



Crystal Lattice of KCl:

