

Cyclic dienes

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Cyclic dienes are of two types:

- 1) Homoannular dienes
- 2) Heteroannular dienes

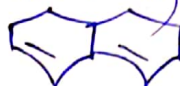
1) Homoannular dienes — conjugated dienes in which the two double bonds are present in the same ring are called homoannular dienes. (Cisoid orientation)

2) Heteroannular dienes — conjugated dienes in which the two double bonds are present in different rings are called heteroannular dienes. (transoid orientation)



Homoannular diene

$\lambda_{max} = 263 \text{ nm}$

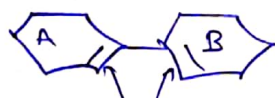


Heteroannular diene

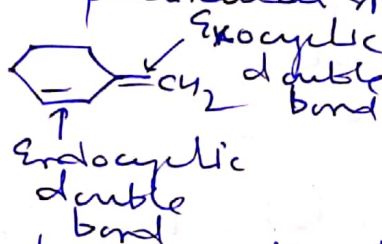
$\lambda_{max} = 234 \text{ nm}$

Effect of endocyclic & exocyclic double bonds on the absorption maxima of dienes

A double bond is said to be endocyclic if it lies within a particular ring and exocyclic if it lies outside a particular ring.



Endocyclic double bonds



Note: In general, each endocyclic double bond increases the ~~double~~ wavelength of absorption of a given cyclic conjugated diene by 5 nm.

WOODWARD-FIESER rules for calculation of λ_{max} of simple conjugated dienes

Woodward formulated certain empirical rules for calculating λ_{max} for the long wavelength $\pi \rightarrow \pi^*$ transition for dienes & polyenes.

These rules further modified by Fieser in 1957. According to these rules, each diene has a certain fixed basic value and the value of λ_{max} depends upon the following points

- 1) The no. of alkyl substituents or ring residues present in the diene system
- 2) The no. of double bonds extending conjugation beyond the diene system

3. The presence of polar groups or auxochromes⁽²⁾
 Such as $-X$ (Cl, Br etc) or $(-OH, NH_2)$ etc

Rules for calculating λ_{max} of the $\pi-\pi^*$ absorption band of dienes

Parent heteroannular or open chain diene 214 nm

Parent homoannular diene 253 nm

for each double bond extending conjugation add 30 nm

for each alkyl substituent or ring residue add 5 nm

for each exocyclic double bond add 5 nm

Auxochromes

$-OCOCH_3$ - add 0 nm

$-OR$ add 6 nm

$-SR$ add 30 nm

$-NR_2$ add 60 nm

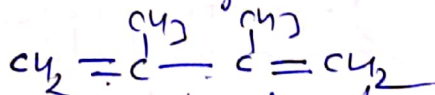
$-Cl, -Br$ add 5 nm

Q1 Calculate the λ_{max} for the most intense band in UV spectrum of

1) 2,3-dimethyl-1,3-butadiene

2) 1,4-dimethylcyclohexa-1,3-diene

1 2,3-dimethyl-1,3-butadiene



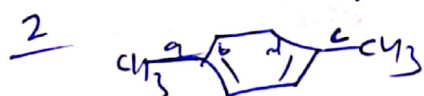
It has two methyl substituents $\Rightarrow$

Parent Value of acyclic diene = 214 nm

Two alkyl substituent = $2 \times 5 = 10$ nm

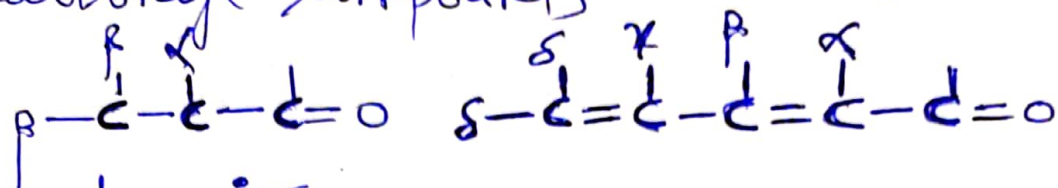
Calculated $\rightarrow \lambda_{max} = 214$

10
224 nm



It is a homoannular diene with the two methyl substituents marked as (a & c) and the two ring residues marked as (b & d)

Woodward-Fieser rules for calculation of λ_{max} for $\pi-\pi^*$ absorption band of α,β -unsaturated carbonyl compounds



Base values:

- 1) α,β -unsaturated acyclic or six membered ring ketone - 215nm
- 2) α,β -unsaturated five membered ring ketone - 202nm
- 3) α,β -unsaturated aldehydes - 207nm
- 4) α,β -unsaturated acids or esters - 197nm

Increments

- 1) Each alkyl group or ring residue

α - 10nm

β - 12nm

γ and higher - 18nm

- 2) Each double bond extending conjugation - 30nm

- 3) Each exocyclic double bond - 5nm

- 4) Homoannular diene component - 39nm

Auxochromes Positions

	Positions	β	γ	δ
-OH	35nm	30nm	30nm	50nm
-OR	35nm	30nm	17nm	31nm
-OCOR	6nm	6nm	6nm	6nm
-Cl	15nm	12nm	—	—
-Br	25nm	30nm	—	—
-NH ₂		95nm		

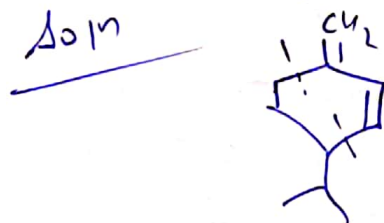
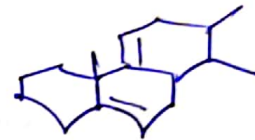
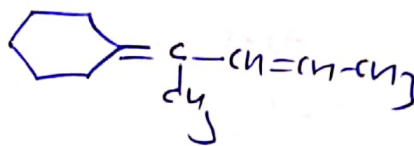
Parent Value of homoannular diene = 253 nm
 Two methyl substituents (2x5) = 10 nm
 Two ring residues (2x5) = 10 nm
 $\lambda_{max} = 273 \text{ nm}$

Q3 Calculate λ_{max} for diene
 Soln The given compound is heteroannular diene with four ring residues

Parent value of heteroannular diene = 214 nm
 four ring residues (4x5) = 20 nm

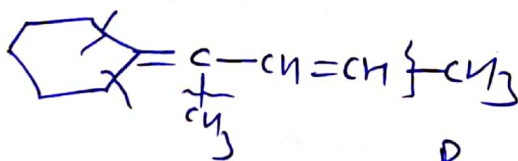
$\lambda_{max} = 234 \text{ nm}$

Q4 Calculate λ_{max} -



Parent Value of cyclic conjugated diene = 214 nm
 Two ring residues = (2x5) nm
 One exocyclic double bond = 5 nm

$\lambda_{max} = 229 \text{ nm}$



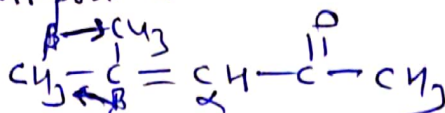
Parent Value = 214 nm
 Two methyl substituent = 10 nm
 Two ring residue = 10 nm
 One exocyclic double bond = 5 nm
 $\lambda_{max} = 239 \text{ nm}$



Parent Value = 253 nm
 One double bond extending conjugation = 30 nm
 five ring residue (5x5) = 25 nm
 Three exocyclic double bond = 15 nm
 $\lambda_{max} = 323 \text{ nm}$

Ques

Calculate λ_{max} for the following compounds (5)



Soln

The given compound is α,β -unsaturated ketone having two alkyl substituents at β -positions

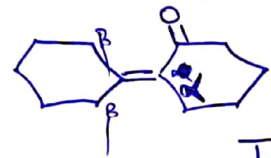
Basic value for α,β -unsaturated acyclic ketone = 215 nm

Two alkyl substituents at β position (2x12) = 24 nm
calculated λ_{max} = 239 nm

Q2

Calculate the λ_{max} Value.

1)



Basic value = 215 nm

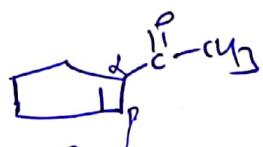
One α -ring residue (1x10) = 10 nm

Two β -ring residues (2x12) = 24 nm

One double bond α -cyclic to two rings (2x5) = 10 nm

259 nm

2)



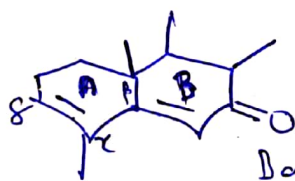
Basic value for α,β -unsaturated acyclic ketone = 215 nm

One α -ring residue = 10 nm

One β -ring residue = 12 nm

237 nm

3)



Basic value of α,β -unsaturated

sin membered ring ketone = 215 nm

One double bond α -cyclic conjugation = 30 nm

One double bond α -cyclic to ring A = 5 nm

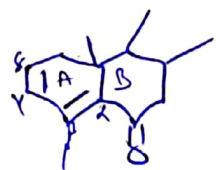
One β -ring residue = 12 nm

One γ -ring residue = 18 nm

One δ -ring residue = 18 nm

298 nm

4)



Basic - 215 nm

One double bond α -cyclic conjugation = 30 nm

One homoannular diene component = 39 nm

One α -ring residue = 10 nm

One δ -ring residue = 18 nm

One double bond α -cyclic to B = 5 nm

Total 377 nm