## SHIVAJI COLLEGE, UNIVERSITY OF DELHI DEPARTMENT OF CHEMISTRY

## **INTERNAL TEST-I (Academic Year 2023-24)**

Name of the Course: B.Sc. (P) Life Sciences Semester: VI (A)

Name of the Paper: Organometallics, Bioinorganic Chemistry, Polynuclear Hydrocarbons and

UV, IR Spectroscopy (Unique Paper Code: 42177926)

Faculty Name: Dr. Sunil Yadav and Mr. Deepesh Singh

**Duration**: 1 Hour **Maximum Marks**: 10 **Date of Test**: 16.04.2024

Attempt any four questions. All questions carry equal marks.

- Calculate the EAN number of HMn(CO)<sub>5</sub> and predict the number of metal- metal bonds in Mn<sub>2</sub>CO<sub>10</sub>.
  (2.5)
- 2. With reference to biological system define essential metal ions, trace metal and non-essential metal ions with an example of each. (2.5)
- 3. Explain the structure of haemoglobin and its function in biological system. (2.5)
- 4. Define organometallic compounds. Classify the organometallic compounds based on the nature of metal carbon bond with an example of each. (2.5)
- 5. Arrange the following species according to decreasing order of the CO stretching frequency and give reasons:
  - (i)  $[V(CO)_6]^-$ ,  $[Cr(CO)_6]$ ,  $[Fe(CO)_6]^{2-}$

(ii) 
$$[Mn(CO)6]^+, [V(CO)6]^-, [Cr(CO)_6]$$
 (2.5)

- 6. Is B(OCH<sub>3</sub>)<sub>3</sub> considered as an organometallic compound? Calculate Effective Atomic Number for the following compounds:
  - (i) Fe<sub>2</sub>(CO)<sub>9</sub>
  - (ii)  $[Mn(CO)_5(\eta^2-C_2H_4)]$

(iii) 
$$[Cr(\eta^6-C_6H_6)(CO)_2]$$
 (2.5)

